

ION-HEAT MODIFICATION OF INORGANIC DIELECTRICS PROPERTIES

A.V.Kabyshev, F.V.Konusov

High Voltage Research Institute at Tomsk Polytechnic University,
Lenin av. 2a, Tomsk, 634050, Russian Federation, phone: (3822) 423870, e-mail: kabyshev@hvd.tpu.r

The influence of an ions irradiation and subsequent annealing in vacuum on mechanical and electrophysical properties of inorganic dielectrics examined. The opportunity of purposeful regulation of their surface resistivity in limits $10^{10} \dots 10^7$ Ohm per squ (Ω/\square) is shown. The modes of modification ensuring deriving on a surface of dielectrics a thermostable resistive coatings are give

Introduction

The irradiation of a solid state by ions calls in a near-surface layer complex of changes affecting nuclear and electronic subsystems. Atoms displacement and pulverization are result of elastic interaction of ions with lattice atoms. As against metals in dielectrics the role of inelastic interactions accompanying with atoms ionization and electronic subsystem excitation is important. The concentration of electronic excitations will increase with diminution of ions mass and with magnification of their energy. Result of elastic and inelastic interaction is the formation in near-surface layer of new disequilibrium state distinguished on structure and properties from initial. Postimplantation annealing, stimulating in near-surface layer the complex physico-chemical and structural - phase changes, calls further transformation of dielectrics properties and fixes the reached changes.

After ion-heat modification the near-surface layer of inorganic dielectrics represents "composition" containing at the same time with defects and implanted elements unregulated distribution of conductive and - or semiconductive particles having different size and shape [1]. The majority of particles is separated by dielectric layers with high concentration of defects and thickness $1.0 \dots 10^4$ nm. There are also chains of contacting among themselves particles. The marked building elements and inhomogeneities differently influence on materials properties.

In this work the brief analysis of literary dates including mechanical properties change of dielectrics irradiated with continuous ions bunches by a current density $1 \dots 10 \mu\text{A}/\text{cm}^2$ is adduced. The authors investigations of ion-heat modification of electrophysical properties of materials are given. In both cases the single crystals of oxides and alkali-halide compounds, oxide and nitride ceramics of different composition utilised. The irradiation of materials for investigation of electrophysical properties was fulfilled in a pulse-frequency mode by ions at energy $50 \dots 150$ keV, the current density in impulse made $10^2 \dots 10^3$ A/cm², the fluence varied in limits $\Phi = 10^{15} \dots 10^{17}$ cm⁻². The post-implantation annealing was carried out in vacuum at residual pressure $P < 1$ Pa and $T_{\text{an}} = 300 \dots 2000$ K.

I. Mechanical properties change

The ions implantation influences on strengthening properties of near-surface layer and last effects on the properties of volume. Microhardness [2-7] and stability to crack formation at local loading [2-4,6,8] and

strength of adhesion of dielectrics are raise [9,10]. Coefficient of friction and the wear rate reduced [2,3,11,12]. The flexural strength is incremented $10 \dots 30\%$ on the average [3,6,8,13]. Changes of sin crystals are more significant, than of polycrystals and glasses [3,13-16]. The quantitative changes of properties are defined by conditions of irradiation and subsequent thermal treatment, are stipulated by induced defects generation, their annealing and transformation into complexes, and structural changes, formation of solid solutions and new phases. The contribution of strengthening from dispersion mechanism, solid solution formation and grain - boundaries participation to determining value for doped by ion implantation near surface layer. Deformation strengthening is determined by dislocation structure of underlying layer. The dominant mechanism of strengthening depends on chemical nature of the introduced atoms too. The noble gases ions are applied expediently to modification of mechanical properties after preliminary deposition of metal film on dielectric (film thickness does not exceed depth of ions run) [5,17].

II. Modification of electrophysical properties

The electrophysical properties of dielectrics surface are sensitive to radiative and subsequent thermal ac-

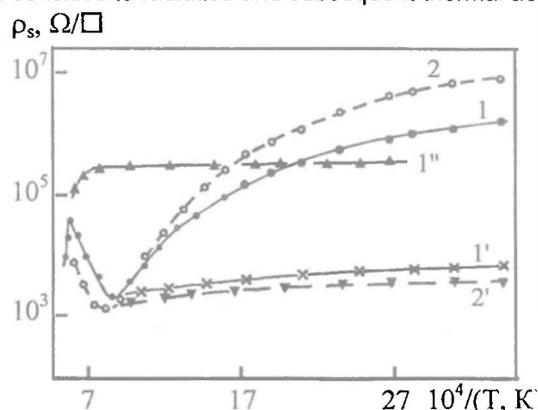


Fig. 1. Surface resistivity of implanted (1, 2) and annealed (1', 2') ceramics on a basis Al_2O_3 : 1, 1', 1''-BK94-2, 1'- after annealing at 1100 K, 1''- after annealing at 1650 K; 2, 2'- BK94-1, 2'- after annealing at 1100 K. A fluence of carbon ions 10^{17} cm⁻².

tions also. The ions irradiation reduces surface resistivity ρ_s on 4...8 orders of magnitude (fig. 1). Reestablishment of irradiated surface up to metal realize

compounds easily losing the anions at ion bombardment (for example, alkali-halides). The enrichment of near-surface layer by component having high pressure of vapours above than some concentration can lead to it spontaneous desorption both at irradiation, and at the subsequent heat treatment in vacuum. Result of desorption is the drop of electrical conduction σ (fig. 2).

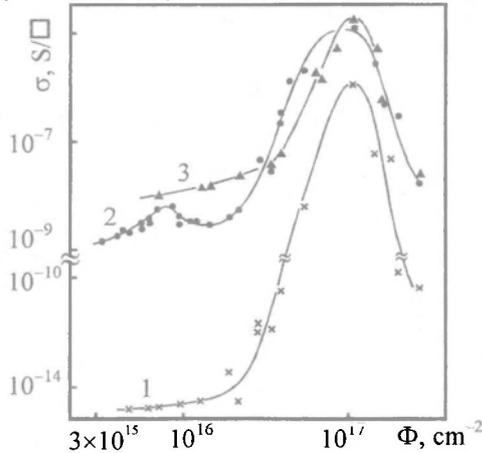


Fig. 2. Effect of carbon ions fluence on an electrical conduction KBr at temperature 290 (1), 650 (2), 720 K (3).

The postimplantation heat treatment promotes the further drop ρ_s (fig. 1). Stabilization $\rho_s(T)$ of heat-resistant materials occurs after annealing at 950...1500 K. Result of ion-heat modification is forming on surface a thermostable conductive coating. Surface resistivity can be governed by treatment modes in limits $10^3 \dots 10^{15} \Omega/\square$. The temperature coefficient of ρ_s does not exceed $10^{-4} \dots 10^{-3} \text{ K}^{-1}$ in temperature interval up to 1500 K. An optimum range of annealing is individual for each material and makes

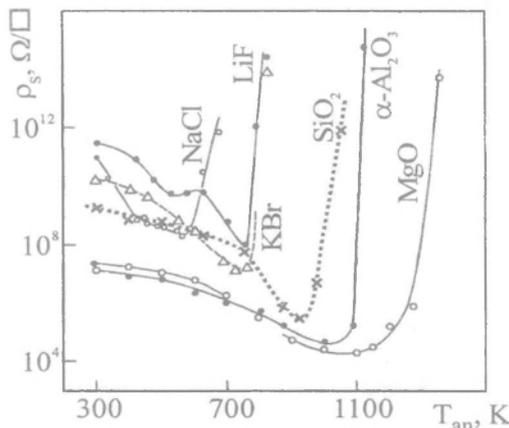


Fig. 3. Postimplantation annealing temperature effect on change of surface resistivity at $T=300$ K of single crystals and quartz glass irradiated $\Phi=10^{17} \text{ C}^+/\text{cm}^2$.

550 ... 750 K for alkali-halide crystals, 950...1200 K for oxide ceramics (fig. 3) and 1100...1500 K for nitride ceramics (fig. 4). Medium of annealing are vacuum or noble gases. This excludes an electronic exchange

between the modified surface and active gas phase reagents calling restoration of properties owing to electronic-ionic reactions between defects.

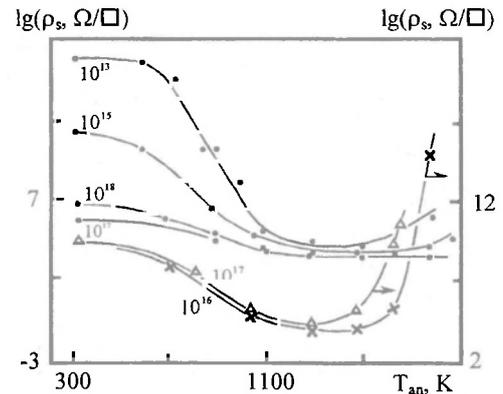


Fig. 4. Postimplantation annealing temperature effect on change of surface resistivity at $T=300$ K irradiated nitride ceramics. Numerals at curves indicates fluence of carbon ions, cm^{-2} : \bullet - boron nitride; Δ - silicon nitride; \times - aluminium nitride.

The ions of light element and ions of elements, capable to substitute lattice atoms and to form semiconductive solid solutions, increment conduction on 10...12 orders of magnitude. Action of heavy ions is less effectively.

The greatest change of dielectric properties is observed at ions fluence $10^{15} \dots 10^{17} \text{ cm}^{-2}$ in materials having high portion of covalent component in interatomic bond (nitride ceramics, fig. 4). Mode of ionic intermixing allowing owing to introduction of recoil atoms and radiation-enhanced diffusion to increase additionally concentration of impurity elements in modified layer is effectively too. Essential changes of conduction of multicomponent oxides systems (where ionic component of chemical bond dominates) are observed only in materials with low contents of SiO_2 both as glasses phase in intercrystallite boundaries and as basic crystalline phase. The conductive coating formed on surface of nitride ceramic materials is more thermostable than a similar coating on oxide ceramic.

The coating on polycrystals is more stable against high temperature effect, than analogous coating on single crystals and glasses (compare fig. 3 and 4). It is stipulated by presence in polycrystals high concentration of boundaries. Boundaries and near-boundary regions of crystallites, having distorted electronic structure and uncompensated charge, determine a spatial distribution, sink and fixing component of Frenkel pairs at irradiation, and degree of transformation defects clusters and diffusion of impurity at heating too. Segregation of new phases at annealing are fixed at first on boundaries, and at higher temperatures in crystallites also. The stability of formed compounds in polycrystalline materials is higher, than in single crystals. Such situation is exhibited especially explicitly on nitride ceramics having a composite hierarchy structure with the fragments size 50...100 nm. Heat stability of any modified materials properties raises, if

implanted elements are formed with matrix a stable covalent bonds.

Influence of chemical interaction between introduced elements and matrix atoms at postimplantation annealing observed on example of polycrystalline corundum irradiated by ions Cu^{+n} and Ti^{+n} , having fraction substitutional cations of lattice Al_2O_3 accordingly 0.05 and 0.9 [18]. The modified materials differ not only behaviour $\rho_s(T_{\text{an}})$ and thermal stability, but also mechanism of conduction. Positive sign of temperature coefficient of conduction (diminution σ with T growth up to 800...900 K) is characteristic only for polycrystalline corundum irradiated by Ti^{+n} ions and annealed in vacuum at 1400...1700 K. The investigations of electronic structure and chemical reactions have shown, that for system $\text{Ti-Al}_2\text{O}_3$ characteristically strong chemical interaction of titanium with oxygen and aluminium, and for system $\text{Cu-Al}_2\text{O}_3$ only with oxygen [19].

In alkali-halide crystals the most essential properties changes are called by ionic intermixing mode (fig. 5). Their conduction at 300 K by modes of ion-heat

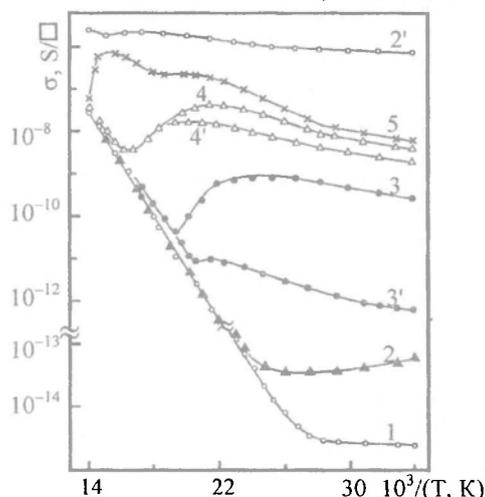


Fig. 5. Electrical conduction of crystals LiF after ions implantation (1-5), postimplantation heat treatment at 500 K (3', 4') and modification in ionic intermixing mode (2'). A fluence of carbon ions, cm^{-2} : 1- 1.6×10^{16} ; 2, 2'- 3.8×10^{16} ; 3, 3'- 5.5×10^{16} ; 4, 4'- 7.7×10^{16} ; 5- 1.4×10^{17} .

modification can be enlarged up to $10^5 \dots 10^6$ Siemens per square (S/\square) (fig. 3, 5). Conductive near-surface layer has both positive, and negative temperature coefficient of conduction. The conduction growth is stipulated by transformation of the induced defects in impurity-vacancy complexes. Diminution σ is result of changes of concentration and morphology of metal formations (these formations are formed in volume and on a surface of modified layer) and their removal from surface at $T=530 \dots 600$ K. Annealing of defects at $T > 700$ K also reduces σ . The enrichment of modified layer by metal realizes owing to selective pulverization of material at irradiation and diffusion of F-centres to surface at heating. Coatings on alkali-halide crystals surface having stability in time conduction $\sigma > 10^{-7}$ S/\square at $T > 380$ K are formed at modification only in ionic in-

termixing mode (fig. 5). The activation energy charge carriers in such layers makes 0.2...0.5 eV.

Ions implantation in ceramics beforehand heat up to $T_i > 500$ K and subsequent annealing do not all to create a modified layer with $\sigma > 10^{-8}$ S/\square . Stable against temperature action properties are formed at annealing at 900...1200 K ($T_i < 1000$ K) and/or 1700...1750 K ($T_i > 500$ K). Conduction of n-type donors in layers having $\sigma > 10^{-9}$ S/\square , however shift Fermi level to conduction band is expressed to lesser degree, than in case of strong magnification conduction ($T_i < 500$ K, $\sigma \geq 10^{-4}$ S/\square). The conduction p-type is characteristic for a weak-conductive surface having $\sigma \leq 10^{-11}$ S/\square . The behaviour $\rho_s(T)$ is determined by a stability to T_i of generated point defects and their simple complexes. The defects, induced at high temperature implantation, are much more stable, than defects proper low-temperature implantation.

Conclusion

The ion-heat action modifies surface properties inorganic dielectrics. Semiconductive coating formed by modification is thermally stable and can be utilized for practical purposes at temperature up to 1800 K.

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