

**Magnetic and photoluminescent properties
of $\text{La}_{1-x}\text{Nd}_x\text{InO}_3$ solid solutions ($0.007 \leq x \leq 0.05$)
with perovskite structure**

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For $\text{La}_{1-x}\text{Nd}_x\text{InO}_3$ ($x = 0.007, 0.02, 0.05$), $\text{La}_{1-x}\text{Nd}_x\text{In}_{1-y}\text{Cr}_y\text{O}_3$, $\text{La}_{1-x}\text{Nd}_x\text{In}_{1-y}\text{Mn}_y\text{O}_3$ ($x = 0.05$, $y = 0.005$) solid solutions magnetic susceptibility in the temperature range of 5–300 K in a magnetic field of 0.86 T was measured and effective magnetic moment of Nd^{3+} ions was calculated (Fig. a). Magnetization dependences of these solid solutions on the magnetic field up to 14 T were investigated at 5 K and 300 K.

X-ray diffraction patterns showed that the samples were single-phase and had the structure of orthorhombically distorted perovskite.

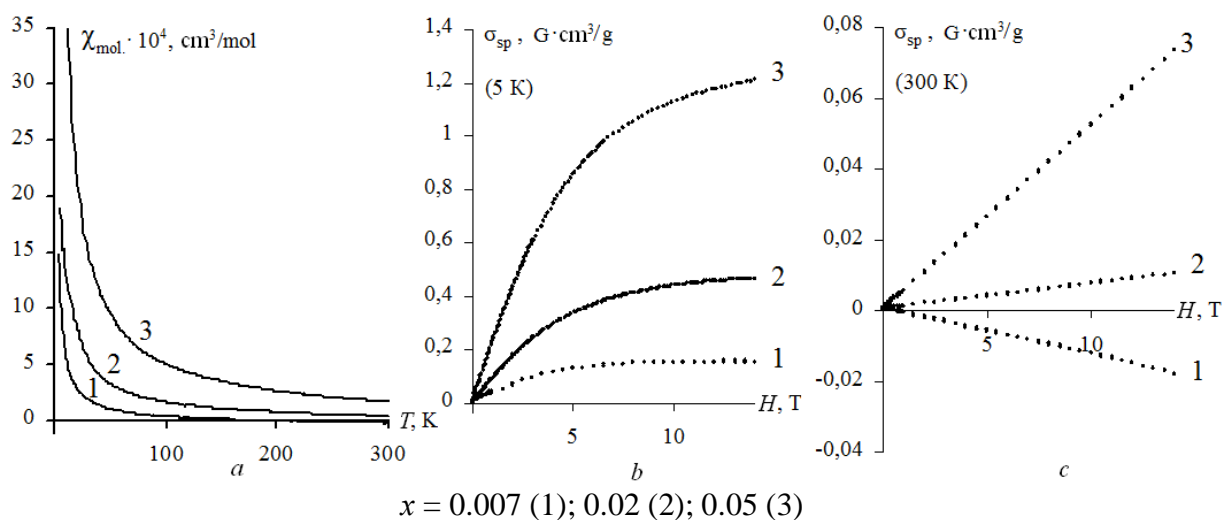


Fig. 1 Magnetic properties of $\text{La}_{1-x}\text{Nd}_x\text{InO}_3$ solid solutions: a – the temperature dependence of the molar magnetic susceptibility, b, c – the field dependence of the specific magnetization at 5 K and 300 K

The effective magnetic moment of the Nd^{3+} ion for $\text{La}_{1-x}\text{Nd}_x\text{InO}_3$ solid solutions with $x = 0.007, 0.02, 0.05$ was significantly smaller than its theoretical value ($3.62 \mu_B$). The magnetic moments of Nd^{3+} ions for $\text{La}_{1-x}\text{Nd}_x\text{InO}_3$ solid solutions with $x = 0.007, 0.02, 0.05$ calculated from their magnetization value at 5 K in a field of 14 T (Fig. b) were equal to $1.18 \mu_B, 1.26 \mu_B, 1.31 \mu_B$, respectively.

With an increase in Nd^{3+} ions content in $\text{La}_{1-x}\text{Nd}_x\text{InO}_3$ from $x = 0.007$ to 0.05 there was a significant increase in photoluminescence intensity at infrared wavelengths. The intensity of the photoluminescence band at $\lambda = 860\text{--}960$ nm for $\text{La}_{0.95}\text{Nd}_{0.05}\text{In}_{0.995}\text{Cr}_{0.005}\text{O}_3$ sample was higher than that for $\text{La}_{0.95}\text{Nd}_{0.05}\text{InO}_3$ sample due to the superposition of Nd^{3+} and Cr^{3+} ions photoluminescence bands.