## PROPERTIES OF ROTATIONAL BANDS OF ISOTOPES Yb

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Present paper focuses on low-lying states of positive parity of isotopes  $^{170,172,174}$ Yb. The calculation is conducted by utilizing a phenomenological model [1] which accounts Coriolis mixture all of the experimentally known low-lying rotational bands states with  $K^{\pi} < 3^{+}$ . Experimentally observed K-forbidden transitions as well as non-adiabaticities of energy and in ratios of E2- transitions can be explained by Coriolis mixture states.

The calculations have been carried out for the  $^{170,172,174}$ Yb isotopes. All experimentally known rotational bands of positive parity with  $K^{\pi} < 3^{+}$  have been included in basis states of Hamiltonian.

The reduced probability of E2-transitions and reduced probability of M1- transitions from the states  $I_iK_i$  to the level  $I_fK_f$  band are calculated. The reduced probabilities of E2 – transitions for <sup>172</sup>Yb are presented in Tables 1.

The experiment suggests that m=5 band with  $K^{\pi}=0_m^+$ , one band  $\ell=1$  with  $K^{\pi}=2_{\ell}^+$ , and  $\nu=19$  with  $K^{\pi}=1_{\nu}^+$  states in  $^{170}\text{Yb}$  [2]. These all  $n=m+\ell+\nu=25$  rotational bands have been included in the basis states of Hamiltonian (1). For the isotopes  $^{172,174}\text{Yb}$ , basis states of Hamiltonian include  $n=m+\ell+\nu=15$  (m=5,  $\ell=2$  and  $\nu=8$ ) and  $n=m+\ell+\nu=22$  (m=5,  $\ell=2$  and  $\nu=15$ ), correspondingly [3,4,5].

The energy and structure of wave functions of excited states are calculated. The reduced probabilities of E2- and M1- transitions are also calculated and comprised with experimental data which are gives satisfactory result.

Table 1. Reduced probability of $E2$ – transitions in the $^{172}$ Yb					
$I_i K_i \rightarrow I_f 0_f$	Exp. [6]	Theory	$I_i K_i \rightarrow I_f 0_f$	Exp.[4]	Theory
$22_{1} \rightarrow 00_{1}$	74.6(57)	82	$20_{2} \rightarrow 00_{1}$	14(1)	13
→ 20 <sub>1</sub>	121 (12)	130	→ 20 <sub>1</sub>	45 (7)	23
→ 40 <sub>1</sub>	6.8 (7)	8.6	→ 40 <sub>1</sub>	142 (20)	74
$32_{\scriptscriptstyle 1} \rightarrow 20_{\scriptscriptstyle 1}$	152 (11)	154	$20_3 \rightarrow 00_1$	0.4(1)	3.6
→ 40 <sub>1</sub>	79 (6)	73	→ 20 <sub>1</sub>	0.6 (4)	3.0

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